

Development of magnetocaloric materials in room temperature magnetic refrigeration application in recent six years

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Abstract: Some magnetocaloric materials were used successfully in magnetic refrigeration application and became one of the critical parts of magnetic refrigeration technology whose delightful progresses were made worldwide in the past 30 years. At the same time, the research on giant magnetocaloric materials will accelerate the development of room temperature magnetic refrigeration. In this paper, the new theoretical and experimental investigations on magnetic materials in room temperature application were described, including Gd and its binary and ternary intermetallic compounds, Mn-based compounds, $\text{La}(\text{Fe}_{13-x}\text{M}_x)$ -based compounds and manganites. Based on the analysis of hysteresis, corrosion, cost and heat process, the comparison between different families of magnetic materials was discussed. Further research of room temperature magnetic refrigerant was suggested.

Key words: magnetic material; refrigeration; magnetocaloric effect; magnetic resistance

1 Introduction

Some magnetic materials, such as lanthanide metal Gd, MnAs, and lanthanide transition-metal-based compounds, generate magnetocaloric effect (MCE). Therefore, these magnetic materials can be adopted in magnetic refrigeration application. Gadolinium (Gd), is the mainly used magnetocaloric material at present. It was firstly applied in room temperature magnetic refrigeration by BROWN in 1976 and then its magnetocaloric physical properties were widely focused. Up to now, the magnetic materials with MCE have become one of the critical parts of magnetic refrigeration.

To improve efficiency of energy utilization and protect environment, investigators in the world have explored new refrigeration technology, as well as new substituting refrigerants. Room temperature magnetic refrigeration is an environment-safe refrigeration technology with many excellent features, such as compact configuration, low noise, high efficiency, high stability and longevity. The newly designed magnetic refrigeration components and systems use water-based heat transfer fluids, and these environmentally desirable products make minimal contributions to global warming. Moreover, efficiency improvements of 20%–30% compared with those of currently available vapor compression-based systems, are envisioned once technology development is completed [1]. Many investigators in America, China, Japan, and some

European countries have focused on the performance research of room temperature magnetic refrigeration since the 1990s because of its great applicable prospect. As a result, room temperature magnetic refrigeration has been one of the hot topics in the world. Currently, it has been regarded as one of the nine advanced subjects of the International Institute of Refrigeration (IIR) and a corresponding international workgroup has been built. However, the room temperature magnetic refrigeration still stays in the stage of experimental investigations, and many practical and critical problems for this refrigeration technology are to be solved. The development of room temperature magnetic refrigerator and its application need further theoretical and experimental investigations on magnetic refrigerant, magnetic cycle, active magnetic regenerator (AMR) and corresponding system.

In 1999, PECHARSKY and GSCHNEIDNER [2] published a review paper describing the phenomenon of the MCE in magnetic materials under different temperature ranges and discussing the relationship between MCE and order-disorder magnetic phase transition. Then some important review articles on the development of magnetocaloric materials have been published in worldwide range since 2000. In 2003, YU et al [3] presented in detail the development of magnetic materials in room temperature regions, including Gd and its alloys, perovskite and perovskite-like compounds, transition metal compounds and composite material. In 2005, GSCHNEIDNER et al [4] reviewed systematically different families of magnetic materials, such as the lanthanide (R) Laves phases (RM_2 , where $\text{M}=\text{Al, Co}$ and

Ni), $Gd_5(Si_{1-x}Ge_x)_4$, $Mn(As_{1-x}Sb_x)$, $MnFe(P_{1-x}As_x)$, $La(Fe_{13-x}Si_x)$ and their hydrides and the manganites ($R_{1-x}M_xMnO_3$, where R=lanthanide and M= Ca, Sr and Ba). In 2007, PHAN and YU [5] reviewed a new class of magnetocaloric materials in the temperature range of 100–375 K, that is, the ferromagnetic perovskite manganites ($R_{1-x}M_xMnO_3$, where R=La, Nd, Pr, and M= Ca, Sr, Ba), and analyzed the nature of these materials. BRÜCK et al [6] reviewed a new class of magnetic refrigerant materials for room temperature applications in conjunction with a magnetic phase transition of the first order, including $Gd_5(Ge,Si)_4$ and related compounds, $La(Fe,Si)_{13}$ and related compounds, MnAs-based compounds, Heusler alloys and Fe_2P -based compounds. Then in 2008, GSCHNEIDNER and PECHARSKY [7] discussed the large scale production of giant magnetocaloric effect materials and potential problems for the use of giant magnetocaloric effect materials in magnetic refrigerators with the recent market investigation on rare earth metal. Based on the new published papers on room temperature magnetic refrigeration in recent six years, and proceedings of the first and the second IIF-IIR international conferences on magnetic refrigeration at room temperature, we presented the nature of MCE and giant magnetocaloric effect (GMCE), the criteria to select magnetic refrigerant, the new development of magnetic material in room temperature region, the comparison between different families of magnetic materials, and the further research of room temperature magnetic refrigerant in this paper.

2 Magnetocaloric effect

The magnetocaloric effect, which is intrinsic to all magnetic materials, is a consequence of magneto-thermal couplings between the magnetic entropy and the lattice entropy. When a magnetic field is applied to the material, the atomic magnetic moments of paramagnetic or soft ferromagnetic materials become aligned, making the material more ordered. Consequently, the materials expel heat and their magnetic entropy decreases. Otherwise, the magnetic moments return to their random directions, and the materials absorb heat and their magnetic entropy increases when the magnetic field is reduced isothermally.

The MCE can be related to the isothermal magnetic entropy change (ΔS_M) and the adiabatic temperature change (ΔT_{ad}), which are shown in the following expression [8–9]:

$$\Delta S_M = \mu_0 \int_{H_0}^{H_i} \left(\frac{\partial M}{\partial T} \right)_H dH \quad (1)$$

$$\Delta T_{ad} = -\mu_0 \int_{H_0}^{H_i} \frac{T}{C_H} \left(\frac{\partial M}{\partial T} \right)_H dH \quad (2)$$

where μ_0 is the permeability of vacuum; H_0 and H_i are the initial and the final magnetic field strength, respectively; C_H is the heat capacity in constant magnetic field; and $(\partial M/\partial T)_{H_0}$ is the derivative of magnetization with respect to temperature in a constant magnetic field.

Other parameters for comparing magnetic materials are the refrigerant capacity, q , which indicates that how much heat can be transferred between the cold and hot sinks in one ideal refrigeration cycle, and the relative cooling power (P_{RCP}). q and P_{RCP} can be established as follows:

$$q = \int_{T_0}^{T_i} \Delta S_M dT \quad (3)$$

$$P_{RCP}(S) = -\Delta T_{FWHM} \Delta S_M(T, H) \quad (4)$$

$$P_{RCP}(T) = -\Delta T_{FWHM} \Delta T_{ad}(T, H) \quad (5)$$

where ΔT_{FWHM} is the full-width at half-maximum of ΔS_M or ΔT_{ad} .

3 Criteria for selecting room temperature magnetic material

On the basis of the corresponding theoretical analysis and the nature of MCE, magnetic materials in magnetic refrigeration should satisfy several features for application, including [3, 5, 10]: (1) the large ΔS_M and ΔT_{ad} (i.e. large total angular momentum number (J) and Lande factor (g) for ferromagnetic material); (2) the large density of magnetic entropy, which is an important factor contributing to the working efficiency of materials; (3) the small lattice entropy (i.e. the high Debye temperature); (4) the modest Curie temperature (T_C) in the vicinity of room temperature to guarantee that the large magnetic entropy change can be obtained in the whole temperature range of the cycle; (5) the nearly zero magnetic hysteresis; (6) the very small thermal hysteresis; (7) the small specific heat and large thermal conductivity, which ensure remarkable temperature change and rapid heat exchange; (8) the large electric resistance (i.e. the lowering eddy current heating or the small eddy current loss); (9) the high chemical stability and simple sample synthetic route.

4 Giant magnetocaloric effect material

The discovery of the GMCE is a landmark in the development of room temperature magnetic refrigeration [11]. The GMCE results from the fact that a structural transformation takes place simultaneously with the magnetic ordering. In this case, both the magnetic sub-lattice and crystallographic sub-lattice are easily affected by the magnetic field. Therefore, the MCE in the

first-order phase transition materials undergoing the coupled magneto-structural transformations (or the GMCE compounds) arises from the added difference of the entropies of the two crystallographic modifications (polymorphs) of the material [8]. Further research on the GMCE shall focus on its nature and mechanism.

For the GMCE materials, the added entropy change resulted from structural transformation in the first-order phase transition can be expressed by ΔS_{st} , and hence the total entropy change is given by

$$\Delta S_M = \mu_0 \int_{H_0}^{H_i} \left(\frac{\partial M}{\partial T} \right)_H dH + \Delta S_{st} \quad (6)$$

An estimation based on comparison of the MCEs exhibited by closely related materials with and without the magnetic field-induced structural transformation indicates that ΔS_{st} may account for more than a half of the total magnetic entropy change in magnetic fields less than 5 T [8, 12].

Fig.1 schematically compares the magnetocaloric effects in the first-order magnetic phase transition (FOMT) compounds (GMCE materials) and the second-order magnetic phase transition (SOMT) compounds (conventional MCE materials) [8], where T_C is the transition temperature. It can be seen from the figure that the GMCE materials have much larger magnetic entropy changes compared with those of the conventional MCE compounds at any temperature, and the former have much wider applicable prospect.

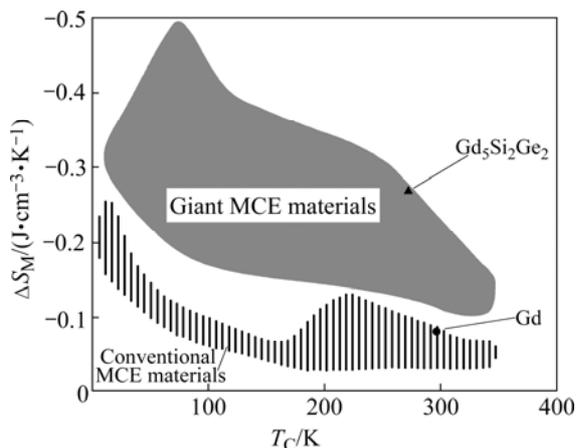


Fig.1 Comparison between GMCE materials and conventional MCE materials at $\mu_0 \Delta H = 5$ T [8]

5 New research progress of room temperature magnetic material

The new research development of MCE materials near room temperature mainly focuses on Gd and its binary and ternary intermetallic compounds, Mn-based compounds, $\text{La}(\text{Fe}_{13-x}\text{M}_x)$ -based compounds and manganites, which are to be reviewed in the next four

sections.

5.1 Gd and its binary and ternary intermetallic compounds

As the mainly applied magnetic material at present, Gd has large MCE at the ordering temperature of $T_C = 294$ K. Fig.2 illustrates the maximum MCE of Gd measured indirectly or directly at its ordering temperature under different magnetic field strength [8]. The solid drawn through the data points represents the power-law fit of the data. It can be seen from the figure that the maximum ΔT_{ad} of Gd is in the range of 2.3–6.0 K for magnetic field changes between 0.5 and 2.0 T in permanent magnets.

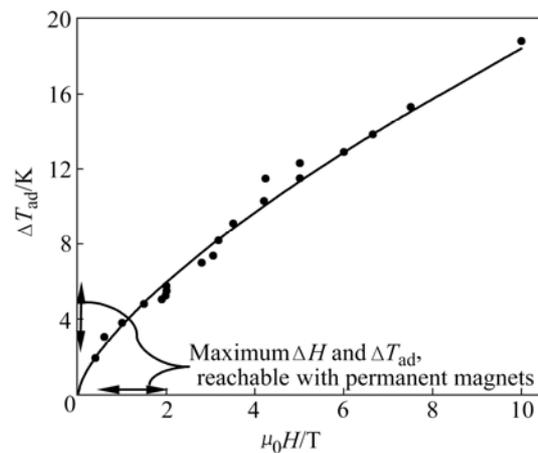


Fig.2 The maximum MCE of Gd at its ordering temperature (Fitted equation of MCE for Gd at T_C : $\Delta T_{ad} = 3.675 \times (\mu_0 H)^{0.7}$) [8]

GSCHNEIDNER et al [4] summarized the binary and ternary intermetallic compounds studied recently, including RCO_2 -based systems ($R = \text{Dy, Ho, and Er}$) [13–16], RAl_2 -based systems ($R = \text{Pr, Nd, Tb, Dy, Ho, Er, and Tm}$) [17], RNi_2 -based systems ($R = \text{Pr, Nd, Gd, Tb, Dy, Ho, and Er}$) [18] and miscellaneous intermetallic compounds [19–20]. The magnetic entropy change and Curie temperature of several magnetic material families at $\Delta H = 5$ T are shown in Fig.3, indicating that T_C of the most binary and ternary intermetallic compounds deviates from room temperature seriously. However, T_C of 13 kinds of compounds, including $\text{Gd}_4(\text{Bi}_{2.25}\text{Sb}_{0.75})$, $\text{Gd}_4(\text{Bi}_{1.5}\text{Sb}_{1.5})$, $\text{Gd}_4(\text{Bi}_{0.75}\text{Sb}_{2.25})$, Gd_4Sb_3 , Gd_4Bi_3 , $\text{Gd}_5\text{Si}_{2.3}\text{Ge}_{1.7}$, $\text{Gd}_5\text{Si}_{2.5}\text{Ge}_{1.5}$, $\text{Gd}_5\text{Si}_3\text{Ge}$, $\text{Gd}_5\text{Si}_{1.98}\text{Ge}_{2.02}$, $\text{Gd}_5\text{Si}_2\text{Ge}_2$, $\text{Gd}_5\text{Si}_{2.09}\text{Ge}_{1.91}$, $\text{Gd}_5\text{Si}_{2.02}\text{Ge}_{1.98}$, and $\text{Gd}_5\text{Si}_{2.1}\text{Ge}_{1.9}$, locates in the near room temperature region, and MCE of the former 8 kinds of refrigerant is smaller than that of Gd with the reverse situation for the latter 5 kinds of materials [4].

In 1997, the GMCE was first found in Gd-Si-Ge (GSG) alloys by PECHARSKY and GSCHNEIDER [21] at the AMES Laboratory, and the ΔS_M and ΔT_{ad} of GSG

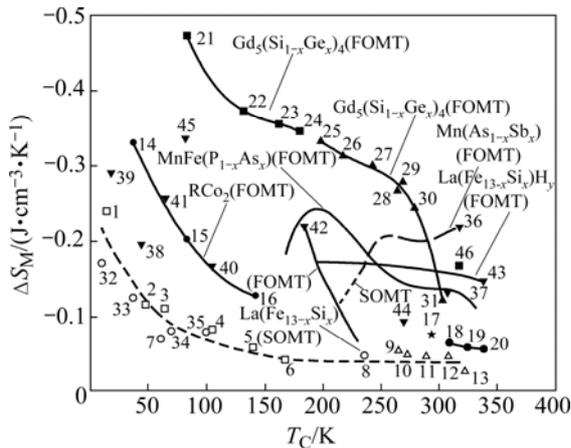


Fig.3 ΔS_M at $\Delta H=5$ T for RCo_2 , RAl_2 , $Gd_5(Si_{1-x}Ge_x)_4$, $Mn(As_{1-x}Sb_x)$, $MnFe(P_{1-x}As_x)$ and $La(Fe_{13-x}Si_x)_y$ families plus a number of individual compounds versus Curie temperature [4]: 1— $ErAl_2$; 2— $(Dy_{0.7}Er_{0.3})Al_2$; 3— $DyAl_2$; 4— $TbAl_2$; 5— $(Tb_{0.4}Gd_{0.6})Al_2$; 6— $GdAl_2$; 7— $Er(Co_{0.85}Si_{0.15})_2$; 8— $TbCo_2$; 9— Gd_4Bi_3 ; 10— $Gd_4(Bi_{2.25}Sb_{0.75})$; 11— $Gd_4(Bi_{1.5}Sb_{1.5})$; 12— $Gd_4(Bi_{0.75}Sb_{2.25})$; 13— Gd_4Sb_3 ; 14— $ErCo_2$; 15— $HoCo_2$; 16— $DyCo_2$; 17— Gd ; 18— $Gd_5Si_2.3Ge_{1.7}$; 19— Gd_5Si_3Ge ; 20— Gd_5Si_4 ; 21— $Gd_5Si_{0.5}Ge_{3.5}$; 22— Gd_5SiGe_3 ; 23— $Gd_5Si_{1.2}Ge_{2.8}$; 24— $Gd_5Si_{1.3}Ge_{2.7}$; 25— $Gd_5Si_{1.5}Ge_{2.5}$; 26— $Gd_5Si_{1.6}Ge_{2.4}$; 27— $Gd_5Si_{1.8}Ge_{2.2}$; 28— $Gd_5Si_{1.95}Ge_{2.05}$; 29— $Gd_5Si_{1.98}Ge_{2.02}$; 30— $Gd_5Si_{2.02}Ge_{1.98}$; 31— $Gd_5Si_{2.1}Ge_{1.9}$; 32— $HoCoAl$; 33— $DyCoAl$; 34— $TbCoAl$; 35— $GdCoAl$; 36— $MnAs$; 37— $MnFeP_{0.45}As_{0.55}$; 38— TbN ; 39— HoN ; 40— $Tb_5Si_2Ge_2$; 41— Dy_5Si_3Ge ; 42— $La(Fe_{11.7}Si_{1.3})$; 43— $La(Fe_{11.5}Si_{1.5})H_{1.8}$; 44— $La_{1.4}Ca_{1.6}Mn_2O_7$; 45— Gd_5Sn_4 ; 46— $Ni_{55.2}Mn_{18.6}Ga_{26.2}$

alloys are larger than those of Gd by 70%–80% [21]. About 140 papers were published on the R_5T_4 materials, where R = a rare-earth element and T = Si, Ge or Sn [4] in mid-2004. The excitement for these compounds is not only due to their GMCE, but can also be ascribed to a number of other unusual features, such as a colossal magnetostriction and giant magnetoresistance. These excellent behaviors are a consequence of a coupled magnetic-structural first-order transition from either the paramagnetic (PM) monoclinic $Gd_5(Si_2Ge_2)$ -type structure to the ferromagnetic (FM) orthorhombic Gd_5Si_4 -type structure, or from the antiferromagnetic (AFM) Sm_5Ge_4 -type structure to the ferromagnetic orthorhombic Gd_5Si_4 -type structure. An interesting feature is that the first-order magnetic phase transition (FOMT) temperature in $Gd_5(Si_{1-x}Ge_x)_4$ is strongly dependent on the molar ratio Si to Ge, as shown in Fig.4 [4]. In the figure, the solid lines indicate the magnetic phase boundaries, and the vertical dotted lines delineate the regions where the alloys are single phase materials.

YUCEL et al [22] investigated the effect of alloying $Gd_5(Si_{2-z}Ge_{2-R_{2z}})$ compound with 3d (Co and Mn) and p (Ga and B) orbitals. They found that the substitution of Si and Ge could adjust T_C of the compound between 20 and

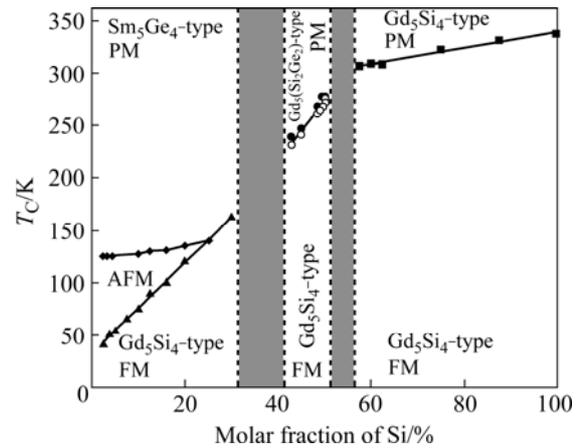


Fig.4 Magnetic phase diagram of Gd_5Si_4 - Ge_5Ge_4 pseudo-binary system [4]: \circ — Monoclinic $Gd_5(Si_2Ge_2)$ -type phase (Heat treated); \bullet — Monoclinic $Gd_5(Si_2Ge_2)$ -type phase (As cast); \blacksquare — Orthorhombic Gd_5Si_4 -type phase; \blacktriangle , \blacklozenge — Orthorhombic Gd_5Ge_4 -type phase

286 K, with ΔS_M variation in this process. For $R=Mn$, the maximum ΔS_M is decreased by up to 17% with the increase of Mn content from $z=0.01$ to 0.1. When R denotes Co and z is above 0.04, the increase of Co content causes strong reduction of T_C down to 260 K with a 13% increase of ΔS_M . For $R=Ga$ or B, the magnetic entropy is slightly reduced by 12% with increasing R content. CHEN et al [11] studied the magnetic properties of $Gd_5Si_{2-x}Ge_{2-x}Sn_{2x}$ ($x=0.05, 0.10, 0.15, 0.20, 0.25$, and 0.50). They reported that ΔS_M of the $Gd_5Si_{2-x}Ge_{2-x}Sn_{2x}$ alloys gradually increases with the increase of Sn content from $x=0$ to $x=0.25$. For $x \leq 0.2$, the alloys have a dominant phase of monoclinic $Gd_5Si_2Ge_2$ -type crystal structure and a small quantity of Gd_5Si_4 -type phase at room temperature, and the latter decreases with the increase of Sn content. The maximum ΔS_M of $Gd_5Si_{1.75}Ge_{1.75}Sn_{0.5}$, which has only one paramagnetic-ferromagnetic magnetic phase transition, is as large as 16.7 J/(kg·K) under a magnetic field change of 1.8 T at its T_C of 269 K. Its MCE exceeds that of Gd about two times. When Dy replaces Gd in Gd_5Si_4 , XIE et al [23] found that T_C decreases in a linear fashion from Gd_5Si_4 ($T_C=338$ K) to Dy_5Si_4 ($T_C=140$ K), but ΔS_M only slightly lowers, by about 8% for $(Gd_{2.5}Dy_{2.5})Si_4$.

The GMCE in Gd-Si-Ge alloys can be negatively influenced by the impurities of the commercial Gd, because the FOMT is destroyed due to the impurities. WU and LU [24] evaluated the MCE of Gd-Si-Ge alloys made of Chinese commercial Gd, in which major impurities are C, O, N, and Ca, and the GMCE is mainly impaired by the first two elements. For magnetic refrigerant Gd, the poor resistance to corrosion in water will seriously impede against its wider application, and then it is needed to examine the influence of physical properties of different heat transfer solutions on

corrosion of magnetic refrigerants. ZHANG et al [25] pointed out that NaOH solution with pH=10 is the best one to inhibit the corrosion of Gd in water.

Table 1 exhibits the MCE of newly investigated Gd-Si-Ge alloys and Mn-based compounds. The conclusion can be drawn from Table 1 that GSG alloys have the modest T_C and large MCE for its application in room temperature regions, and they possess a prominent applicable potential.

5.2 Mn-based compounds

The base material of Mn-based compounds is MnAs. Different alloys, such as Mn(As_{1-x}Sb_x), MnFe(P_{1-x}As_x), and Ni-Mn-Ga Heusler alloys, can be obtained through element substitution [4].

A large value of $\Delta S_M=27$ J/(kg·K) in MnAs can be obtained at a magnetic field change of 6 T. However, MnAs is not suitable for magnetic refrigerant as the FOMT of MnAs is accompanied with large thermal hysteresis. PALACIOS et al [29] pointed out that the problem above can be avoided by partial substitution of Sb for As. For MnAs_{1-x}Sb_x ($0 \leq x \leq 0.4$), ΔS_M reaches a maximum value of 25 J/(kg·K) as the magnetic field change increases from 0 to 6 T, and ΔT_{ad} reaches peak values between 10.8 K at $x=0.4$ and 12.4 K at $x=0$. Moreover, the addition of Sb makes T_C tunable from 317 to 225 K with increasing x from 0 to 0.4, and the peak values of ΔS_M under a magnetic field change of 5 T are in the range of 27–30 J/(kg·K) at $0.05 \leq x \leq 0.30$ [30]. WADA et al [31] noted that Sb stabilizes the NiAs-type structure when $x \geq 0.1$, and the FOMT changes to SOMT, resulting in a reduction of ΔS_M and ΔT_{ad} .

TEGUS et al [32] studied the effect of Cr and Co substitutions for Fe in MnFe(P_{1-x}As_x) alloy. For the Mn(Fe_{1-x}Cr_x)P_{0.47}As_{0.53} alloy, whose T_C is about 305 K, Cr lowers both T_C and ΔS_M . When $\Delta H=2$ T and $x=0.02$, the FOMT changes to an SOMT, and ΔS_M decreases by about 25% at its $T_C \approx 275$ K. For the alloy Mn(Fe_{1-x}Co_x)P_{0.5}As_{0.5}, a 10% substitution of Co for Fe ($x=0.1$) reduces T_C from 282 to 260 K and ΔS_M by about 55% at $\Delta H=2$ T.

Ni-Mn-Ga Heusler alloy is a ferromagnetic alloy that undergoes a reversible FOMT from a tetragonal martensitic phase to the cubic austenitic phase. It has been paid a wide attention due to its large MCE recently [33]. In 2004, CHERECHUKIN et al [34] found a large ΔS_M of 20.7 J/(kg·K) in Ni_{2.18}Mn_{0.82}Ga alloy at a magnetic field change of 1.8 T, which is the best result observed up to present. Then ZHANG et al [33] performed experimental investigations on Ni_{53.6}Mn_{23.2}Ga_{23.2} Heusler alloy. Their results indicate that a plateau in the peak of ΔS_M exists and the distance

between the two peaks is 4 K. ZHOU et al [35] reported large ΔS_M for Ni_{55.2}Mn_{18.6}Ga_{26.2} alloy, with values of 20.37 J/(kg·K) at $\Delta H=5$ T and $T_C=317$ K and 9.46 J/(kg·K) at $\Delta H=1.5$ T. The above description suggests that the Ni-Mn-Ga alloys near the Ni₅₅Mn₁₉Ga₂₆ composition may be good magnetic regenerator alloys for refrigerators/heat pumps operating between 300 and 350 K. However, the ΔS_M peak for Ni_{55.2}Mn_{18.6}Ga_{26.2} is quite sharp, only 5 K wide at half of the peak height maximum, which may limit the use of the Ni-Mn-Ga Heusler alloys as a near room temperature magnetic refrigerant.

Mn contained transition-metal-based compounds also present GMCE. For example, the maximum ΔS_M of 26.1 J/(kg·K) in Mn_{1.1}Fe_{0.9}P_{0.76}Ge_{0.24} alloy at $T_C=299$ K and $\Delta H=5$ T can be achieved [36]. In addition, the series of Fe_{2-x}M_xP (M = Ru, Rh, Pd, Pt) in the Fe₂P type hexagonal structure and R₂Fe₁₇ (R=Pr, Y) system show large value of MCE, and the former shall be explored in further research [37–38].

5.3 La(Fe_{13-x}M_x)-based compounds

The heavy lanthanide metals and their compounds have the largest magnetic moments, which make them the best potential magnetocaloric materials. GUTFLEISCH et al [38] studied the magnetocaloric properties of La(FeSi)₁₃ alloy. They obtained a ΔS_M peak value of 10.3 J/(kg·K) in LaFe_{11.2}Si_{1.8} alloy at 5 T and 240 K, and T_C of LaFe_{11.2}Si_{1.8} alloy can be easily adjusted between 200 and 290 K with the increase of Co content. LIU and ALTOUNIAN [39] investigated the La(Fe_{1-z}Co_z)_{11.4}Si_{1.6} series of alloys from $z=0$ to $z=0.10$. They found that Co additions significantly and linearly increase T_C from about 210 K at $z=0$ to about 330 K at $z=0.10$, but ΔS_M decreases from 23.2 J/(kg·K) at $z=0.02$ to 11.6 J/(kg·K) at $z=0.06$ and $\Delta H=7$ T. HUANG et al [40] studied the magnetocaloric properties of LaFe_{11.17}Co_{0.78}Si_{1.05} alloys prepared with low-purity raw materials in a low magnetic field. Their results indicate that proper addition amount of B to the alloys can improve the stability and enhance MCE and T_C .

For La(Fe_xSi_{1-x})₁₃ alloys, their ΔS_M and ΔT_{ad} can keep large maximum values even in a relatively low magnetic field after controlling T_C by the hydrogenation and the partial substitution of Ce or Pr. FUJITA et al [41–45] investigated the MCEs of La_{1-z}Pr_z(Fe_{0.88}Si_{0.12})₁₃, La_{1-z}Ce_z(Fe_{0.88}Si_{0.12})₁₃, and various La(Fe_{13-y}Si_y)H_y samples. They found that the maximum values of ΔS_M , ΔT_{ad} and P_{RCP} in La_{0.5}Pr_{0.5}(Fe_{0.88}Si_{0.12})₁₃ alloy at a magnetic field change of 5 T are 30 J/(kg·K), 11.9 K and 576 J/kg, respectively, and T_C increases up to 324 K by controlling y . For La(Fe_{0.88}Si_{0.12})₁₃H_{1.6} alloy, its T_C is 330

Table 1 MCE of some Gd-Si-Ge alloys and Mn-based compounds

Alloy	Composition	$\Delta H/T$	T_C/K	$\Delta S_M/(J \cdot kg^{-1} \cdot K^{-1})$	Ref.	
$Gd_5Si_{2-z}Ge_{2-z}R_{2z}$	R=Mn	$z=0.010$	5.0	297	6.40	[22]
		$z=0.025$	5.0	297	6.36	
		$z=0.040$	5.0	297	6.30	
		$z=0.050$	5.0	293	6.10	
		$z=0.100$	5.0	297	5.30	
		$z=0.070$	5.0	297	5.00	
	R=Co	$z=0.015$	5.0	297	6.30	
		$z=0.025$	5.0	297	6.10	
		$z=0.065$	5.0	297	7.10	
	R=Ga	$z=0.010$	5.0	297	6.50	
		$z=0.020$	5.0	297	6.00	
		$z=0.050$	5.0	297	5.60	
	R=B	$z=0.015$	5.0	297	7.00	
		$z=0.025$	5.0	297	6.25	
		$z=0.040$	5.0	297	6.60	
		$z=0.050$	5.0	297	6.90	
$Gd_5Si_{2-x}Ge_{2-x}Sn_{2x}$	$x=0.050$	1.8	308	6.50	[11]	
	$x=0.100$	1.8	294	12.8		
	$x=0.150$	1.8	288	11.4		
	$x=0.200$	1.8	283	12.2		
	$x=0.250$	1.8	269	16.7		
	$x=0.500$	1.8	251	15.9		
$Gd_5Ge_{1.9}Si_2Fe_{0.1}$		5.00	305	7.00	[26]	
$Mn_{1-t}(Ti_{0.5}V_{0.5})_t$	$t=0$	2.00	313.7	68.0	[27]	
		5.00		78.0		
	$t=0.025$	2.00	301.6	15.0		
		5.00		36.0		
	$t=0.050$	2.00	291.0	9.0		
		5.00		22.0		
	$t=0.100$	2.00	266.8	30.0		
		5.00		44.0		
	$t=0.150$	2.00	260.0	23.0		
		5.00		27.0		
	$t=0.200$	2.00	272.6	3.0		
		5.00		7.0		
$Ni_{55.4}Mn_{20.0}Ga_{24.6}$		1.00	313.5	17.4	[28]	
		2.00	313.5	40.1		
		3.00	313.5	60.1		
		4.00	313.5	77.0		
		5.00	313.5	85.8		

K, which is the largest in the $\text{La}(\text{Fe}_{0.88}\text{Si}_{0.12})_{13}\text{H}_y$ compounds. Moreover, it has a large MCE and its refrigerant capacity achieves 130 J/kg, which is almost the same as that of $\text{Gd}_5\text{Ge}_2\text{Si}_2$ and $\text{MnFeAs}_{0.5}\text{P}_{0.5}$ alloy. Fig.5 illustrates the temperature dependence of the isothermal magnetic entropy change for $\text{La}(\text{Fe}_{0.88}\text{Si}_{0.12})_{13}\text{H}_y$ alloys [46]. It can be seen from the figure that the Curie temperature of these compounds can be increased by hydrogen absorption. In addition, the thermal conductivity of $\text{La}(\text{Fe}_x\text{Si}_{1-x})_{13}\text{H}_y$ alloy is better than that of other candidates. Consequently, $\text{La}(\text{Fe}_x\text{Si}_{1-x})_{13}\text{H}_y$ alloy is one of the most practical candidates for magnetic refrigerants working in a wide temperature range. Table 2 lists the magnetocaloric properties of new $\text{La}(\text{Fe}_{13-x}\text{M}_x)$ compounds investigated, indicating that $\text{La}(\text{Fe}_{13-x}\text{M}_x)$ compounds have the modest Curie temperature varying in room temperature region and much larger maximum ΔS_M or P_{RCP} compared with those of Gd through Co addition or hydrogen absorption.

5.4 Manganites

In 1950, JONKER and van SANTEN firstly

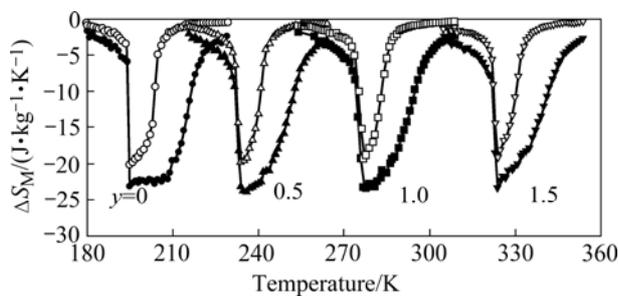


Fig.5 Dependence of isothermal magnetic entropy change ΔS_M on temperature for $\text{La}(\text{Fe}_{0.88}\text{Si}_{0.12})_{13}\text{H}_y$ alloys [46]

synthesized the distorted perovskite-type structure of manganites whose general formula is $\text{A}_{1-x}\text{B}_x\text{MnO}_3$ (where A stands for trivalent rare-earth elements, such as La, Pr, Nd, Sm, Eu, Gd, Ho, Tb, and Y; and B denotes Na^+ , K^+ and Ag^+ , or bivalent alkaline earth ions such as Sr, Ca, Ba, and Pb) [5, 48]. The series of manganites have advantages of low cost, good chemical stability, small coercive force, and large electrical resistivity.

Manganites are now attracting attention in the field of magnetic refrigeration for their large MCE. For example, when the magnetic field changes are 6 and 8 T, the peak values of ΔS_M in $(\text{La}_{0.5}\text{Gd}_{0.2})\text{Sr}_{0.3}\text{MnO}_3$ alloy achieve 7.2 and 8.8 J/(kg·K), respectively [49]. The MCE above can satisfy the magnetic refrigerators presented in recent years. However, T_C of a majority of manganites deviates from room temperature, which will limit their applications in room temperature regions. This problem can be avoided by alloying substituting ions. For example, $\text{La}_{0.70}\text{Sr}_{0.30}\text{MnO}_3$ (LSMO) alloy is a ferromagnet presenting considerable magnetocaloric effect with T_C less than 90 °C, and then T_C lowers effectively to near room temperature when La ion is substituted by Er or Eu ions, with increasing P_{RCP} and keeping constant of ΔS_M in this process [48].

DAS and DEY [50] studied a series of K-doped manganese perovskites of $\text{La}_{1-x}\text{K}_x\text{MnO}_3$ type ($x=0.05, 0.10, \text{ and } 0.15$) with nanometric crystallite size. They noted that T_C of the compound enhances from 260.4 ($x=0.05$) to 309.7 K ($x=0.15$). Increasing K content in the compound increases the maximum ΔS_M at 1 T to 3.00 J/(kg·K), which is about 89% to that reported for Gd. P_{RCP} of LKMO compounds is nearly one-third to that reported for Gd (119 J/kg). HOU et al [51] examined

Table 2 MCE of new $\text{La}(\text{Fe}_{13-x}\text{M}_x)$ compounds studied

Composition	$\Delta H/T$	T_C/K	$\Delta S_M/(\text{J}\cdot\text{kg}^{-1}\cdot\text{K}^{-1})$	$P_{\text{RCP}}/(\text{J}\cdot\text{kg}^{-1})$	Ref.
$\text{La}(\text{Fe}_{11.70}\text{Si}_{1.30})\text{H}_{1.1}$	5.00	287	31.02	–	
$\text{La}(\text{Fe}_{11.57}\text{Si}_{1.43})\text{H}_{1.3}$	5.00	291	28.02	–	[43]
$\text{La}(\text{Fe}_{11.44}\text{Si}_{1.56})\text{H}_{1.0}$	5.00	274	23.03	–	
$\text{La}(\text{Fe}_{11.50}\text{Si}_{1.50})\text{H}_{1.3}$	5.00	288	–	–	[47]
$\text{La}(\text{Fe}_{11.50}\text{Si}_{1.50})\text{H}_{0.9}$	5.00	272	–	–	
$\text{La}(\text{Fe}_{0.88}\text{Si}_{0.12})_{13}\text{H}_{1.0}$	2.00	276	18.80	158	
$\text{La}(\text{Fe}_{0.88}\text{Si}_{0.12})_{13}\text{H}_{1.5}$	2.00	323	19.50	159	[44–45]
$\text{La}(\text{Fe}_{0.90}\text{Si}_{0.10})_{13}\text{H}_{1.0}$	2.00	287	–	173	
$\text{La}(\text{Fe}_{0.90}\text{Si}_{0.10})_{13}\text{H}_{1.1}$	2.00	287	28.00	–	
$\text{LaFe}_{11.17}\text{Co}_{0.78}\text{Si}_{1.05}$	1.48	277	4.00	–	
$\text{LaFe}_{11.07}\text{Co}_{0.78}\text{Si}_{1.05}\text{B}_{0.1}$	1.48	281	4.60	–	[40]
$\text{LaFe}_{10.97}\text{Co}_{0.78}\text{Si}_{1.05}\text{B}_{0.2}$	1.48	287	4.90	–	
$\text{LaFe}_{10.87}\text{Co}_{0.78}\text{Si}_{1.05}\text{B}_{0.3}$	1.48	289	5.50	–	

single-phase polycrystalline samples of $\text{La}_{0.8-x}\text{Nd}_x\text{Na}_{0.2}\text{MnO}_3$ ($x=0, 0.05, 0.10, 0.15$ and 0.20) alloys by the sol-gel method. It is found that T_C decreases with the increase of x , and the maximum ΔS_M , for $x=0.20$, is $0.369 \text{ J}/(\text{mol}\cdot\text{K})$ at about 295 K . HANH et al [52] fabricated $\text{La}_{0.7}\text{Ca}_{0.3-x}\text{Pb}_x\text{MnO}_3$ ($x=0.05, 0.10, 0.15$, and 0.20) alloy with $\text{La}_{1-x}\text{Ca}_x\text{MnO}_3$ and $\text{La}_{1-x}\text{Pb}_x\text{MnO}_3$ by solid-state reaction technique with the expectation that they could establish large MCE in the room temperature region.

Some reports present that the crystal structure plays an important role in the transport, magnetic properties and thermodynamic properties of magnetic refrigerants. However, TERASHITA et al [53] observed the little influence of structural transition, which is in close proximity to the ferromagnetic transition, on ΔS_M in $\text{La}_{1-x}\text{Sr}_x\text{MnO}_3$ compositions. In order to elucidate the effect of lattice structure on the MCE, KIM et al [54] performed an experimental study on ΔS_M in $\text{La}_{0.7}\text{Ca}_{0.3-x}\text{Sr}_x\text{MnO}_3$ ($x=0.120, 0.135$ and 0.150) ferromagnetics. The samples of $x=0.120$ and 0.150 show the usual M-T behavior with T_C of 300 and 323 K , respectively. The composition with $x=0.135$ displays two-step dependence with inflection points at 309 and 320 K , which correspond to T_C of orthorhombic and rhombohedral phases, respectively. Moreover, the maximum values of ΔS_M are $1.87, 1.72$, and $1.7 \text{ J}/(\text{kg}\cdot\text{K})$ for the samples of $x=0.120, 0.135$, and 0.150 , respectively. BEJAR et al [55] discussed new perovskite manganites $\text{La}_{0.7}\text{Ca}_{0.3-x}\text{K}_x\text{MnO}_3$ ($0 \leq x \leq 0.150$). The results show that these compounds crystallize in the orthorhombic structure for $x \leq 0.075$ and become rhombohedral for $x=0.100$ and 0.150 . The disorder for the orthorhombic phase is enhanced with the increase of x content, which is followed by an increase of T_C . For the rhombohedral structure, the disorder is suppressed, and thus T_C decreases.

Table 3 shows the magnetocaloric properties of newly investigated manganites in near room temperature region. From Table 3, it can be noted that T_C can be easily tuned in the temperature range of $290\text{--}360 \text{ K}$ by selecting suitable manganites. The manganite materials with the superior magnetocaloric properties in addition to cheap materials processing cost will be an option of magnetic material in the future magnetic refrigeration technology.

5.5 Comparison between various magnetic refrigerants near room temperature

Many investigators [71–78] revealed that the magnetic hysteresis and thermal hysteresis might have negative influence on magnetic refrigerants exhibiting

the GMCE. However, the greater concern is not the hysteresis of the FOMTs, but the time dependence of ΔT_{ad} [4]. Unfortunately, nothing is known about the time dependence of ΔT_{ad} for the manganites, the $\text{Mn}(\text{As}_{1-x}\text{Sb}_x)$, $\text{MnFe}(\text{P}_{1-x}\text{As}_x)$ and Heusler alloy families.

Moreover, corrosion of different magnetic refrigerants in water-based solution should not be ignored in most magnetic refrigerators. Thus, suitable magnetic materials should be selected on the basis of the factors as mentioned above for given magnetic refrigerator. Table 4 [4] summarizes the advantages and disadvantages of candidate magnetic materials, where a zero indicates that the factor is essentially the same as that of Gd; a plus means that the behaviour is somewhat better than that of Gd, and two pluses mean that the behavior is much better; a minus sign indicates that the property is inferior to that of Gd, and two or three minus signs indicate the behaviour is much worse than that for Gd; the question mark indicates that it is not clear whose specific performance between some magnetic materials and Gd is better. From the table, it can be found that Gd and its solid-solution alloys may occupy their prominent positions in room temperature magnetic refrigeration because of their excellent over-all properties; and LaFeSi alloys may be a suitable choice for their low raw material cost and relatively high MCE in large scale application.

6 Conclusions

(1) Magnetic material is a critical part in magnetic refrigeration, but the present magnetic materials in room temperature region cannot match the demand of practical application. For example, Gd and Gd-Si-Ge alloys have a narrow applicable temperature region, and the Curie temperature of many manganites deviates from room temperature. In order to seek magnetic materials of low cost and large MCE, further studies will be focused on making full use of magnetic entropy change, overcoming magnetic hysteresis and thermal hysteresis, achieving large MCE in relative low magnetic field, and reducing heat capacity of materials under zero magnetic field strength.

(2) The GMCE results from the fact that a structural transformation takes place simultaneously with the magnetic ordering. The next investigation should be performed on the mechanism and relationship between the structural transformation and the magnetic ordering. In other words, the connection between the MCE and the magnetic-phase transformation is to be emphasized in the future study on GMCE materials. The most promising compounds for room temperature magnetic refrigeration

Table 3 Magnetocaloric properties of manganites near room temperature

Composition	Formula	$\Delta H/T$	T_C/K	$\Delta S_M/(J \cdot kg^{-1} \cdot K^{-1})$	$P_{RCP}/(J \cdot kg^{-1})$	Ref.
(La-Sr)MnO ₃	La _{0.845} Sr _{0.155} MnO ₃	1.35	310	1.72	61	[56]
	La _{0.815} Sr _{0.185} MnO ₃	7.00	280	7.10	533	[57]
	La _{0.800} Sr _{0.200} MnO ₃	7.00	305	7.90	395	[57]
	La _{0.650} Sr _{0.350} MnO ₃	1.00	305	2.12	106	[58]
(La-Ba)MnO ₃	La _{0.7} Ba _{0.3} MnO _{3.00}	1.00	336	1.60	36	[59]
	La _{2/3} Ba _{1/3} MnO _{3.00}	1.00	337	2.70	68	[60]
	La _{2/3} Ba _{1/3} MnO _{2.98}	1.00	312	2.60	65	[60]
	La _{2/3} Ba _{1/3} MnO _{2.95}	1.00	300	2.55	69	[60]
	La _{2/3} Ba _{1/3} MnO _{2.92}	1.00	275	1.80	90	[60]
	La _{2/3} Ba _{1/3} MnO _{2.90}	1.00	268	1.70	94	[60]
(La-K)MnO ₃	La _{0.95} K _{0.05} MnO ₃	1.00	260	2.73	282 ^a	[50]
	La _{0.90} K _{0.10} MnO ₃	1.00	287	2.74	286 ^a	[50]
	La _{0.85} K _{0.15} MnO ₃	1.00	310	3.00	303 ^a	[50]
(La-Pb)MnO ₃	La _{0.8} Pb _{0.2} MnO ₃	1.35	310	1.30	–	[61]
	La _{0.8} Pb _{0.2} MnO ₃	1.50	294	1.22	92	[62]
(La-Ca-Sr)MnO ₃	La _{0.70} Ca _{0.25} Sr _{0.05} MnO ₃	5.00	275	10.5	462	[63]
		1.50	315	2.23	–	[64]
	La _{0.70} Ca _{0.20} Sr _{0.10} MnO ₃	2.00	315	2.85	–	[64]
		5.00	308	7.45	374	[63]
		2.00	315	0.88	–	[65]
	La _{0.50} Ca _{0.30} Sr _{0.20} MnO ₃	2.00	317	1.52 ^b	–	[65]
(La-Eu-Sr)MnO ₃	La _{0.49} Eu _{0.21} Sr _{0.30} MnO ₃	1.00	312	1.28	–	[48]
	La _{0.56} Eu _{0.14} Sr _{0.30} MnO ₃	1.00	308	1.42	–	[48]
(La-Ca-Ba)MnO ₃	La _{0.70} Ca _{0.18} Ba _{0.12} MnO ₃	1.00	298	1.85	45	[66]
	La _{0.70} Ca _{0.06} Ba _{0.24} MnO ₃	1.00	320	1.72	44	[66]
(La-Ca-Pb)MnO ₃	La _{0.60} Ca _{0.30} Pb _{0.10} MnO ₃	1.35	289	2.55	56	[67]
	La _{0.70} Ca _{0.25} Pb _{0.05} MnO ₃	1.35	270	3.72	113	[52]
	La _{0.70} Ca _{0.20} Pb _{0.10} MnO ₃	1.35	295	2.53	76	[52], [67]
	La _{0.70} Ca _{0.15} Pb _{0.15} MnO ₃	1.35	317	2.59	80	[52]
	La _{0.70} Ca _{0.10} Pb _{0.20} MnO ₃	1.35	337	3.72	71	[52], [67]
(La-Ca-K)MnO ₃	La _{0.7} Ca _{0.250} K _{0.050} MnO ₃	2.00	270	3.95	88	
		3.00	270	4.92	138	
	La _{0.7} Ca _{0.225} K _{0.075} MnO ₃	2.00	281	3.75	81	[68–69]
		3.00	280	4.72	131	
	La _{0.7} Ca _{0.100} K _{0.100} MnO ₃	2.00	272	3.49	74	
		3.00	273	4.49	121	
(La-Nd-Ba)MnO ₃	La _{0.65} Nd _{0.05} Ba _{0.30} MnO ₃	1.00	325	1.57	24	
	La _{0.63} Nd _{0.07} Ba _{0.30} MnO ₃	1.00	307	1.59	26	[70]
	La _{0.60} Nd _{0.10} Ba _{0.30} MnO ₃	1.00	285	1.85	27	
	La _{0.55} Nd _{0.15} Ba _{0.30} MnO ₃	1.00	269	2.22	31	
(La-Nd-Na)MnO ₃	La _{0.8} Na _{0.2} MnO ₃	1.00	334	0.430 8 ^c	–	[51]
	La _{0.75} Nd _{0.05} Na _{0.2} MnO ₃	1.00	322	0.430 4 ^c	–	[51]
	La _{0.70} Nd _{0.10} Na _{0.2} MnO ₃	1.00	312	0.469 1 ^c	–	[51]
	La _{0.65} Nd _{0.15} Na _{0.2} MnO ₃	1.00	301	0.302 8 ^c	–	[51]
	La _{0.60} Nd _{0.20} Na _{0.2} MnO ₃	1.00	295	0.369 2 ^c	–	[51]
	Gd	5.00	294	10.2	410	[5]
Gd ₅ Si ₂ Ge ₂	5.00	276	18.4	535	[5]	

Note: a—unit is mJ/cm³; b—Heat processing; c—Unit is J/(mol·K).

Table 4 Advantages and disadvantages of various magnetic refrigerants

Factor	Gd	Gd ₅ T ₄	RMnO ₃	LaFeSi	MnAs	FeMnPAs	Ni ₂ MnGa
Raw material cost	0	–	++	++	++	++	+
Preparation	0	–	--	--	--	--	--
Vapor Pressure	0	0	0	0	--	---	0
Fabrication(sheet)	0	–	–	–	–	–	–
MCE, ΔS_M	0	++	–	+	+	+	+
MCE, ΔT_{ad}	0	+	–	–	–	–	?
Refrigerating capacity	0	+	?	+	?	+	?
Hysteresis	0	--	0 ^a	–	–	–	–
Time dependence of ΔT_{ad}	0	–	?	–	?	?	?
Environmental concern	0	0	0	0	–	–	0
Corrosion	0	++	?	–	?	?	?

Note: “a” represents that for SOMT manganites is zero; and for the few FOMT manganites the zero becomes a minus.

include: Gd₅(Si₂Ge_{4-x}) alloys with x near 2; MnAs and MnAs_{1-x}Sb_x; FeMn(P_{1-x}As_x) and related compounds; and La(Fe_{0.88}Si_{0.12})₁₃H_y with $y \geq 1$.

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